# Evidence for cytochrome $bc_1$ complex involvement in nitrite reduction in a photodenitrifier, *Rhodopseudomonas sphaeroides* forma sp. denitrificans

# Katsuro Urata and Toshio Satoh

Department of Biology, Faculty of Science, Tokyo Metropolitan University, Fukazawa 2-1-1, Setagaya-ku, Tokyo 158, Japan

#### Received 19 March 1984

The cytochrome  $bc_1$  complex which has a quinol: cytochrome  $c_2$  oxidoreductase activity was isolated from a photodenitrifier, *Rhodopseudomonas sphaeroides* forma sp. *denitrificans*. When duroquinol (tetramethyl-p-benzoquinol) was used as the electron donor, maximum nitrite reduction activity was obtained in an anaerobic mixture containing the isolated cytochrome  $bc_1$  complex, nitrite reductase and cytochrome  $c_2$ . Nitrite reduction in chromatophore membranes, as well as in the reconstituted system, was inhibited in the presence of antimycin A. No nitrite reduction occurred in the absence of cytochrome  $c_2$ . These results indicated that the cytochrome  $bc_1$  complex involved in photosynthetic electron transfer also donates reducing equivalents to the dissimilatory nitrite reductase via cytochrome  $c_2$  in this photodenitrifier.

Cytochrome bc1 complex

Dissimilatory nitrite reduction Rhodopseudomonas sphaeroides

Denitrification

#### 1. INTRODUCTION

A photosynthetic cyclic electron transfer chain of *Rhodopseudomonas sphaeroides* consists of a ubiquinol:cytochrome  $c_2$  oxidoreductase (a cytochrome  $bc_1$  complex), a photochemical reaction center complex and cytochrome  $c_2$  [1]. Recently, the cytochrome  $bc_1$  complex was isolated from R. sphaeroides [2-4]. It contains cytochromes b and  $c_1$ , iron-sulfur protein and ubiquinone as the essential redox components and is analogous to the mitochondrial cytochrome  $bc_1$  complex in terms of components and function [5].

A photodenitrifier, R. sphaeroides forma sp. denitrificans, can reduce nitrate to nitrogen as an energy-transforming reaction other than oxygen respiration and photosynthesis [6]. The nitrite reductase involved in denitrification uses reduced cytochrome  $c_2$  as the immediate electron donor [7] and a succinate-reduced minus nitrite-oxidized difference spectrum indicates contribution of both b-

and c-type cytochromes in nitrite reduction [8]. From these findings, we have assumed that denitrifying and photosynthetic electron transfer chains share a common electron transfer chain, the cytochrome  $bc_1$  complex and cytochrome  $c_2$  in this photodenitrifier. Recently, participation of the cytochrome  $bc_1$  complex in anaerobic dissimilatory nitrite reduction in denitrification of *Paracoccus denitrificans* was also suggested from evidence that the nitrite reduction is inhibited by antimycin A [9,10].

Here, we confirmed the above prediction by reconstituting the nitrite-reducing system with the cytochrome  $bc_1$  complex, nitrite reductase and cytochrome  $c_2$ .

# 2. MATERIALS AND METHODS

A green mutant strain of R. sphaeroides forma sp. denitrificans IL-106 [8] was grown under denitrifying conditions in light as in [7].

Chromatophore membranes were prepared by sonication as in [11].

The cytochrome  $bc_1$  complex was isolated essentially as in [3] except that the duration of cholate/deoxycholate treatment was shortened to 2 h to obtain higher activity. The final preparations in 50 mM Tris-HCl (pH 8.0), containing 0.05% sodium cholate and 20% glycerol (v/v) were frozen in liquid nitrogen and stored at  $-20^{\circ}$ C.

Nitrite reductase was purified as in [7] and stored in the presence of 50% glycerol (v/v) at  $-20^{\circ}$ C. Cytochrome  $c_2$  was isolated as in [12].

Nitrite reduction activity was assayed at  $25^{\circ}$ C under argon by determining nitrite consumed in the reaction medium as in [13]. The reaction was initiated by adding  $10 \,\mu l$  of 25 mM duroquinol and terminated by adding  $20 \,\mu l$  saturated zinc acetate after 20 min incubation. The activity of cytochrome  $c_2$  reduction was measured spectrophotometrically at 550 nm. Cytochrome content was determined from the reduced minus oxidized difference spectrum with  $\epsilon_{\rm mM}$  values of 20 and 19.1 at the peaks of the  $\alpha$ -band for b- and c-type cytochromes, respectively [14].

Potentiometric titrations and data analyses were carried out as in [11]. The cytochrome  $bc_1$  complex in 3 ml of 50 mM Tris-HCl (pH 8.0)/0.05% sodium cholate in an anaerobic stirred cuvette corresponded to 0.5  $\mu$ M cytochrome  $c_1$ . In titration of cytochrome  $c_2$ , it was 0.5  $\mu$ M in 50 mM Tris-HCl (pH 8.0).

Duroquinol was prepared from duroquinone (Tokyo Chemical Industry Co.) as in [15]. DDC (N,N-diethyldithiocarbamate) and antimycin A were purchased from Wako and Sigma, respectively.

### 3. RESULTS AND DISCUSSION

The isolated cytochrome  $bc_1$  complex contained both b- and c-type cytochromes whose  $\alpha$ -bands were at 559 and 552 nm, respectively. Potentiometric titrations revealed only 3 cytochrome components (table 1). The c-type cytochrome was titrated as a single component with an  $E_m$  value of +248 mV which agrees well with that for cytochrome  $c_1$  [3] and is distinct from that for cytochrome  $c_2$  (+309 mV). About equal amounts of two b-type cytochromes were found with  $E_m$ 

Table 1

Potentiometric titrations of cytochromes in the isolated cytochrome bc<sub>1</sub> complex from Rhodopseudomonas sphaeroides forma sp. denitrificans

Cytochrome	$E_{ m m8.0}$ (mV)	Relative content (%)	
b-type	- 56	30	
	+ 86	25	
c-type	+ 248	45	
$c_2$	+ 309	-	

The relative contents of cytochromes in the cytochrome  $bc_1$  complex are given as percent of the total cytochrome content. The redox potential of cytochrome  $c_2$  was obtained as a control experiment by titrating the purified cytochrome  $c_2$  in 50 mM Tris-HCl (pH 8.0)

values of -56 and +86 mV compatible with those reported in [3]. The ratio between b- and c-type cytochromes was almost 1:1 and compatible with that reported in [4], whereas in [2,3] it was reported that the ratio of b- and c-type cytochromes was 2. The isolated cytochrome  $bc_1$ complex catalyzed cytochrome  $c_2$  reduction with duroquinol as the electron donor, although duroquinol also reduced non-enzymatically cytochrome  $c_2$ . The rate of enzymatic electron transfer through the cytochrome  $bc_1$  complex to cytochrome  $c_2$  was calculated to be 3.71  $\mu$ mol cytochrome  $c_2 \cdot$ nmol cytochrome  $c_1^{-1} \cdot h^{-1}$ , which is about 60% of that reported in [2]. These results indicated that the isolated cytochrome  $bc_1$  complex is comparable to that of R. sphaeroides with respect to cytochrome composition and cytochrome  $c_2$  reduction activity.

Table 2 shows the nitrite reduction activities in various reaction media. Maximum activity was obtained in the complete reaction medium containing nitrite reductase, cytochrome  $c_2$  and the cytochrome  $bc_1$  complex. The activity in the reaction medium without the cytochrome  $bc_1$  complex was due to non-enzymatic reaction of duroquinol with cytochrome  $c_2$ . Direct electron flow to the nitrite reductase from the cytochrome  $bc_1$  complex did not seem to occur, since the activity without cytochrome  $c_2$  was as low as that without both cytochrome  $c_2$  and the cytochrome  $bc_1$  complex, due to direct reaction of duroquinol with nitrite reductase. These results revealed that the

Table 2

Nitrite reduction activities in the various reconstituted systems

Components of reaction medium	Relative activity (%)
Complete	100
Without the cytochrome $bc_1$ complex	48
Without cytochrome $c_2$ Without the cytochrome $bc_1$ complex	20
and cytochrome $c_2$	13
Without nitrite reductase	0

The complete reaction medium (1 ml) consisted of 50 mM sodium phosphate (pH 7.0), 0.05% sodium cholate,  $32\,\mu\text{M}$  cytochrome  $c_2$ , the cytochrome  $bc_1$  complex (0.15  $\mu\text{M}$  cytochrome  $c_1$ ),  $2.3\,\mu\text{M}$  nitrite reductase, and  $0.13\,\text{mM}$  KNO<sub>2</sub>. The activities are expressed as percent of NO<sub>2</sub> consumption in the complete system, the activity of which was  $2.48\,\text{nmol}$  NO<sub>2</sub> reduced · min<sup>-1</sup>

cytochrome  $bc_1$  complex and cytochrome  $c_2$  took part in dissimilatory nitrite reduction as well as photosynthetic cyclic electron transfer in denitrifying R. sphaeroides.

Chromatophore membranes prepared from cells grown under photodenitrifying conditions contain nitrite reductase [16]. Nitrite reduction in chromatophore membranes, as well as in the reconstituted system, was inhibited in the presence of antimycin A, which is well-known to interrupt electron flow in the cytochrome  $bc_1$  complex (table 3). This also supported the fact that the cytochrome  $bc_1$  complex actually participates in nitrite reduction of the photodenitrifier. In the presence of DDC, an inhibitor of nitrite reductase [7], nitrite reduction in both the reconstituted system and chromatophore membranes was inhibited.

The photodenitrifier was shown by potentiometric titrations to possess the same composition of b-type cytochromes as non-denitrifying R. sphaeroides [11] and the cytochrome  $bc_1$  complex from the photodenitrifier was also shown here to

Table 3

Effect of inhibitors on nitrite and cytochrome  $c_2$  reduction in the reconstituted system and in chromatophore membranes from *Rhodopseudomonas sphaeroides* forma sp. *denitrificans* 

Inhibitor	Relative activity (%)				
	Reconstituted system		Chromatophore membranes		
	Nitrite reduction	Cytochrome c <sub>2</sub> reduction	Nitrite reduction	Cytochrome c <sub>2</sub> reduction	
None	100 <sup>a</sup>	100 <sup>b</sup>	100°	100 <sup>d</sup>	
Antimycin A (50 $\mu$ M)	34	30	12	8	
DDC (100 µM)	12	~	7	-	

<sup>&</sup>lt;sup>a</sup> 5.86 nmol NO<sub>2</sub> · nmol cyt. $c_1^{-1}$  · min<sup>-1</sup>

Reaction medium for cytochrome  $c_2$  reduction consisted of 50 mM potassium phosphate (pH 7.0), 0.05% sodium cholate, 38  $\mu$ M cytochrome  $c_2$ , 0.15  $\mu$ M cytochrome  $bc_1$  complex and 100  $\mu$ M duroquinol. Chromatophore membranes were suspended in 50 mM potassium phosphate (0.29 mg protein/ml) containing 38  $\mu$ M cytochrome  $c_2$  and 100  $\mu$ M duroquinol. Nitrite reduction in the reconstituted system was determined as described in table 2. The reaction medium for chromatophore membranes consisted of 50 mM potassium phosphate, 0.25 mM KNO<sub>2</sub>, chromatophore membranes (0.29 mg protein/ml) and 200  $\mu$ M duroquinol. The activity is corrected for that without the cytochrome  $bc_1$  complex and expressed as percent of that without an inhibitor

<sup>&</sup>lt;sup>b</sup> 3.71  $\mu$ mol cyt. $c_2 \cdot$ nmol cyt. $c_1^{-1} \cdot h^{-1}$ 

c 3.73 nmol NO<sub>2</sub>·mg protein<sup>-1</sup>·min<sup>-1</sup>

<sup>&</sup>lt;sup>d</sup> 0.97  $\mu$ mol cyt. $c_2 \cdot$ mg protein<sup>-1</sup>·h<sup>-1</sup>

be identical to that of non-denitrifying R. sphaeroides with respect to cytochrome components (table 1). Furthermore, the cytochrome  $bc_1$  complex isolated from cells of denitrifying R. sphaeroides grown in light in the absence of nitrate (photosynthetic conditions) gave the same nitrite reduction activity in the reconstituted system as listed in table 2 (not shown). Therefore, as far as the nitrite reduction system is concerned, the difference between the photodenitrifier and the nondenitrifying R. sphaeroides seems to be only in the possession of nitrite reductase itself. It is interesting from the viewpoint of evolution of biological energy transformation that the cytochrome  $bc_1$  complex also takes part in anaerobic respiration in addition to photosynthesis and oxygen respiration.

### **ACKNOWLEDGEMENT**

We wish to thank Dr Keizo Shimada for helpful suggestions and discussions.

## REFERENCES

[1] Dutton, P.L. and Prince, R.C. (1978) in: The Photosynthetic Bacteria (Clayton, R.K. and Sistrom, W.R. eds) pp.525-570, Plenum, New York.

- [2] Gabellini, N., Bowyer, J.R., Hurt, E., Melandri, B.A. and Hauska, G. (1982) Eur. J. Biochem. 126, 105-111.
- [3] Takamiya, K., Doi, M. and Okimatsu, H. (1982) Plant Cell Physiol. 23, 987-997.
- [4] Yu, L. and Yu, C. (1982) Biochem. Biophys. Res. Commun. 108, 1285-1292.
- [5] Matsuura, K. and Dutton, P.L. (1981) in: Chemiosmotic Proton Circuits in Biological Membranes (Skulachev, V.P. and Hinkle, P.C. eds) pp.105-122, Addison-Wesley, Reading, MA.
- [6] Satoh, T., Hoshino, Y. and Kitamura, H. (1976) Arch. Microbiol. 108, 265-269.
- [7] Sawada, E., Satoh, T. and Kitamura, H. (1978)Plant Cell Physiol. 19, 1339-1351.
- [8] Bannai, M. and Satoh, T. (1982) Plant Cell Physiol. 23, 205-211.
- [9] Alefounder, P.R., McCarthy, J.E.G. and Ferguson, S.J. (1981) FEMS Microbiol. Lett. 12, 321–326.
- [10] Alefounder, P.R., Greenfield, A.J., McCarthy, J.E.G. and Ferguson, S.J. (1983) Biochim. Biophys. Acta 724, 20-39.
- [11] Yokota, S., Urata, K. and Satoh, T. (1984) J. Biochem. (Tokyo), in press.
- [12] Bartsch, R.G. (1971) Methods Enzymol. 23, 344-363.
- [13] Nicholas, D.J.D. and Nason, S. (1957) Methods Enzymol. 3, 981-984.
- [14] Hatefi, Y., Haarik, A.G. and Griffiths, D.E. (1962) J. Biol. Chem. 237, 1681-1685.
- [15] Rich, P.R. (1981) Biochim. Biophys. Acta 637, 28-33.
- [16] Sawada, E. and Satoh, T. (1980) Plant Cell Physiol. 21, 205-210.